

# Atom Properties & Forces

Name: Key 2019  
 Date: \_\_\_\_\_ Period: \_\_\_\_\_

Part 1: Atom Properties. *Fill in the blanks for the elements listed below.*

Element	Element Symbol	Number of Protons	Number of Neutrons	Number of Electrons	Mass Number	Atomic Number	Symbol w/ AN and MN
Sodium	Na	11	12	11	23	11	$^{23}_{11}\text{Na}$
Calcium	Ca	20	20	20	40	20	$^{40}_{20}\text{Ca}$
Nitrogen	N	7	6	7	13	7	$^{13}_7\text{N}$
Neon	Ne	10	10	10	20	10	$^{20}_{10}\text{Ne}$
Oxygen	O	8	8	10	16	8	O <sup>-2</sup>
Copper	Cu	29	35	29	64	29	$^{64}_{29}\text{Cu}$
Fluorine	F	9	10	9	19	9	$^{19}_9\text{F}$
Aluminum	Al	13	14	10	27	13	$^{27}_{13}\text{Al}^{+3}$

Part 2: Isotopes. *Fill in the blanks for the elements listed below. Remember not to change atomic number or number of protons for atoms of the same element!!*

Element	Number of Protons	Number of Neutrons	Number of Electrons	Mass Number	Atomic Number	Symbol w/ AN and MN
Carbon-12	6	6	6	12	6	$^{12}_6\text{C}$
Carbon-13	6	7	6	13	6	$^{13}_6\text{C}$
Carbon-14	6	8	6	14	6	$^{14}_6\text{C}$
Chlorine-35	17	18	17	35	17	$^{35}_{17}\text{Cl}$
Chlorine-37	17	20	17	37	17	$^{37}_{17}\text{Cl}$
Argon-36	18	18	17	36	18	$^{36}_{18}\text{Ar}$
Argon-38	18	20	18	38	18	$^{38}_{18}\text{Ar}$
Argon-40	18	22	18	40	18	$^{40}_{18}\text{Ar}$

1. Describe the force that keeps the electrons from leaving the atom.

Electric force keeps the  $e^-$  from leaving the atom.

- acts between charged particles, in this case  $p^+$  &  $e^-$  (opposites attract)

- The nucleus of the atom is positive & attracts the negatively charged electrons.

2. Describe the force which keeps the nucleons bound together.

The strong force holds the  $p^+$  &  $n^0$  together in the nucleus. It is  $100\times$  stronger than the electric force & is extremely short-range.

The strong force attracts  $p^+$  &  $p^+$ ,  $n^0$  &  $n^0$ , and  $p^+$  &  $n^0$ .

3. If the electromagnetic force is not strong enough to keep electrons in the cloud, what happens as a result?

If the electric force isn't strong enough to keep the  $e^-$  in the cloud,  $e^-$  may be lost. This forms a positive ion.

4. If the strong nuclear force is not strong enough to keep the nucleons together, what happens as a result?

If the strong nuclear force isn't strong enough to keep the nucleons together, the nucleus will be unstable & break down (decay).