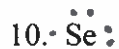
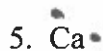
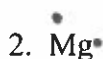


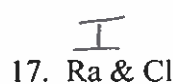
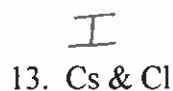
Bonding Practice

Name: Key 2017

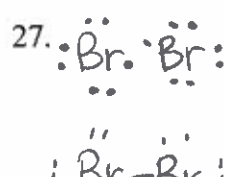
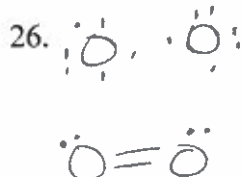
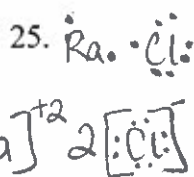
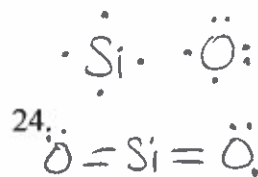
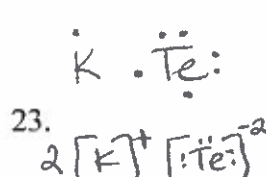
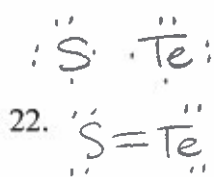
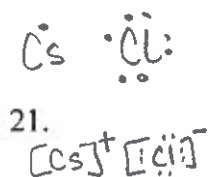
Draw Lewis Dot structures for the following elements.



Ionic bonds are normally formed between metals and non-metals. Covalent bonds are normally formed between two non-metals. State the type of bond that is most likely between the given elements.



For each of the covalent bonds listed above, draw a Lewis Dot structure. For each of the ionic bonds listed above, draw the two ions in the bond.



On the back of this sheet, draw a skeleton periodic table and label everything that you know!

